#### Key Knowledge:

- Nucleic acids as information molecules that encode instructions for the synthesis of proteins
- The structure of DNA, the three main forms of RNA and a comparison of their respective nucleotides
- Amino acids as the monomers of a polypeptide chain and the resultant hierarchical levels of structure that give rise to a functional protein
- Proteins as a diverse group of molecules that collectively make an organism's proteome, including enzymes as catalysts in biochemical pathways

### **NUCLEIC ACIDS**

Nucleic acids are the genetic material of the cell and are composed of recurring monomeric subunits called **nucleotides**. Each nucleotide is comprised of three principal components: pentose sugar, phosphate group and nitrogenous base. Nucleotides combine to form gene sequences that encode specific proteins.

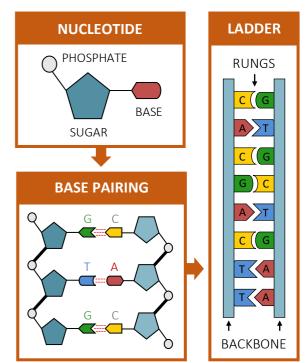
#### DNA

DNA is composed of double-stranded polynucleotide chains. The phosphate and sugar from two separate nucleotides are joined by a **covalent bond** (condensation reaction) to form a sugar-phosphate backbone. Two DNA chains are then held together by **hydrogen bonds** between complementary bases.

DNA has four nitrogenous bases. Adenine and guanine are double-ringed purine molecules, while cytosine and thymine are single-ringed pyrimidine molecules. In terms of pairing:

- Adenine (A) pairs with thymine (T) via two H bonds
- Cytosine (C) pairs with guanine (G) via three H bonds

In order for the complementary bases to pair, the two DNA strands must be running in opposite directions, and hence the two strands are **antiparallel**. Double-stranded DNA will arrange into the most stable configuration: a **double helix**.



### RNA

DNA is a master template while RNA functions as a transient copy. There are three main types of RNA:

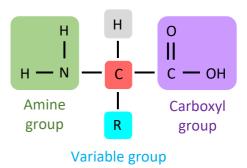
- Messenger RNA (mRNA): A transcript of a DNA sequence that is translated by ribosomes into protein
- **Transfer RNA** (tRNA): Brings amino acids to the ribosome according to the specific mRNA sequence
- Ribosomal RNA (rRNA): A component of the ribosome that enables it to interact with mRNA and tRNA

There are a number of key differences between the structures and functions of DNA and RNA:

- In DNA nucleotides the pentose sugar is deoxyribose, whereas RNA nucleotides use the sugar ribose
- In RNA, the nitrogenous base thymine is replaced by uracil (i.e. DNA uses T, whereas RNA uses U)
- Whereas DNA is a double-stranded molecule, RNA is always a single-stranded molecule
- DNA is confined to the nucleus, whereas RNA molecules enact their functions within the cytosol

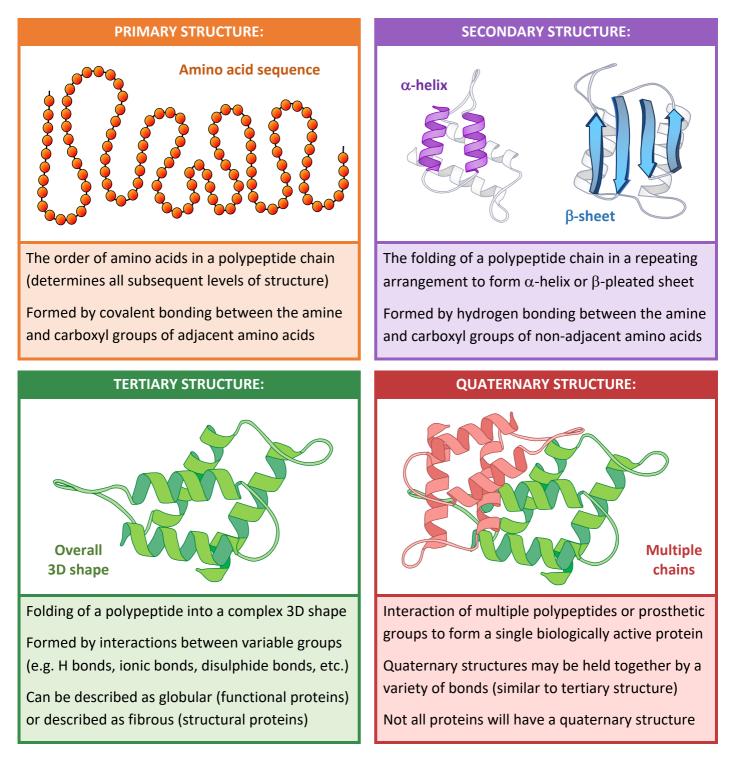
# PROTEINS

Proteins serve a wide variety of functions in living organisms. They enact the genetic instructions encoded by DNA, and thus regulate cellular activity. Proteins are comprised of long chains of recurring monomers called **amino acids**. Amino acids share a common basic structure, with a central carbon atom bound to an **amine group**, a **carboxyl group** and a variable side chain. There are 20 amino acids, each with a distinct side chain (i.e. **R group**) with specific properties.



## **PROTEIN STRUCTURE**

Amino acids are joined via condensation polymerisation to form polypeptides that are linked by peptide bonds. These polypeptide chains may be organised into four hierarchical levels of protein structure:



## DENATURATION

The structure of a protein determines its chemical properties and hence contributes to biological function.

- An  $\alpha$ -helix may increase the **tensile strength** of a protein, while  $\beta$ -sheets confer **mechanical stability**
- Soluble proteins will need their surface to be lined with polar or charged amino acids (hydrophilic), while insoluble proteins will need their surface to be lined with non-polar amino acids (hydrophobic)

**Denaturation** is a change in protein structure that causes a loss of biological activity (usually permanent).

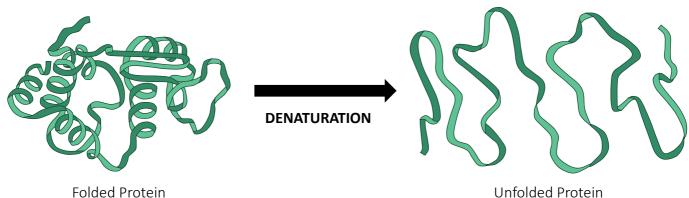
The denaturation of a protein can most typically be caused by two conditions – temperature and pH

#### **Temperature:**

- High levels of thermal energy may disrupt the hydrogen bonds that hold the protein together
- As the bonds are broken, the protein will begin to unfold and lose its capacity to function as intended
- Temperatures at which proteins begin to denature may vary, but most human proteins will optimally function at body temperature (37°C)

#### pH:

- Amino acids are zwitterions, neutral molecules with both negative (COO<sup>-</sup>) and positive (NH<sub>3</sub><sup>+</sup>) regions
- Changing the pH will alter the charge of a protein, which in turn will alter protein solubility and shape
- All proteins have an optimal pH which is dependent on the environment in which it typically functions



**Unfolded** Protein

#### **PROTEIN FUNCTIONS**

Proteins are a very diverse class of organic compounds that serve a wide variety of different roles in cells. Examples of protein functions include:

- **Structure** (e.g. collagen is found in skin, keratin is found in hair)
- **Hormones** (e.g. insulin and glucagon regulate blood sugar levels) .
- **Immunity** (e.g. immunoglobulins target foreign pathogens) ٠
- **Transport** (e.g. protein channels enable facilitated diffusion) ٠
- **Sensation** (e.g. rhodopsin is an eye pigment required for vision) •
- **Movement** (e.g. actin and myosin are used in muscle contraction) •
- **Enzymes** (e.g. Rubisco is responsible for carbon fixation in plants) ٠



Hint: Use a mnemonic device!

The totality of all proteins expressed within a cell, tissue or organism is called the proteome. The proteome of any given individual will be unique, and protein expression levels can change with time and conditions.